

China + Iran = Unreliable?

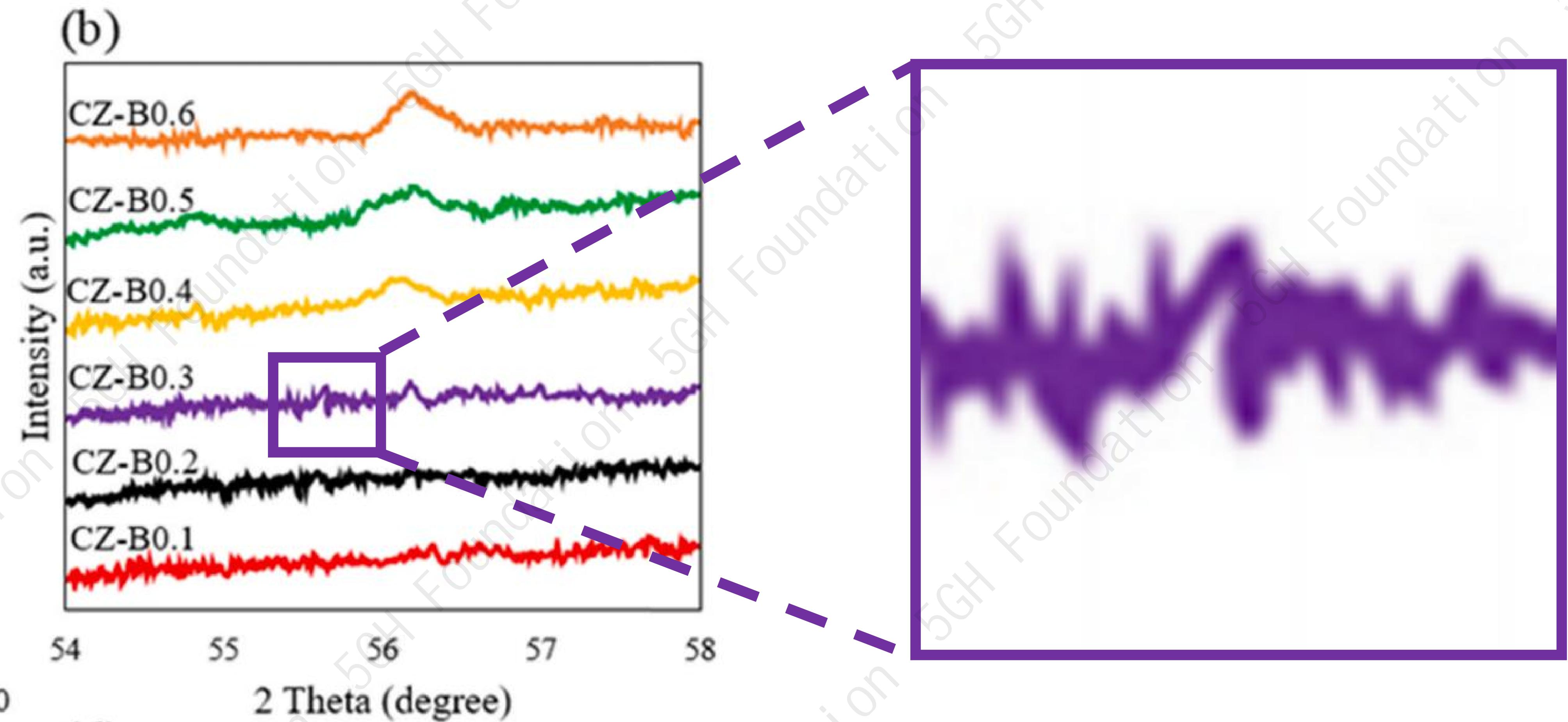
Cross-sectional Study on *Material Chemistry and Physics*

The 5GH Team

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10.1016/j.matchemphys.2024.129428

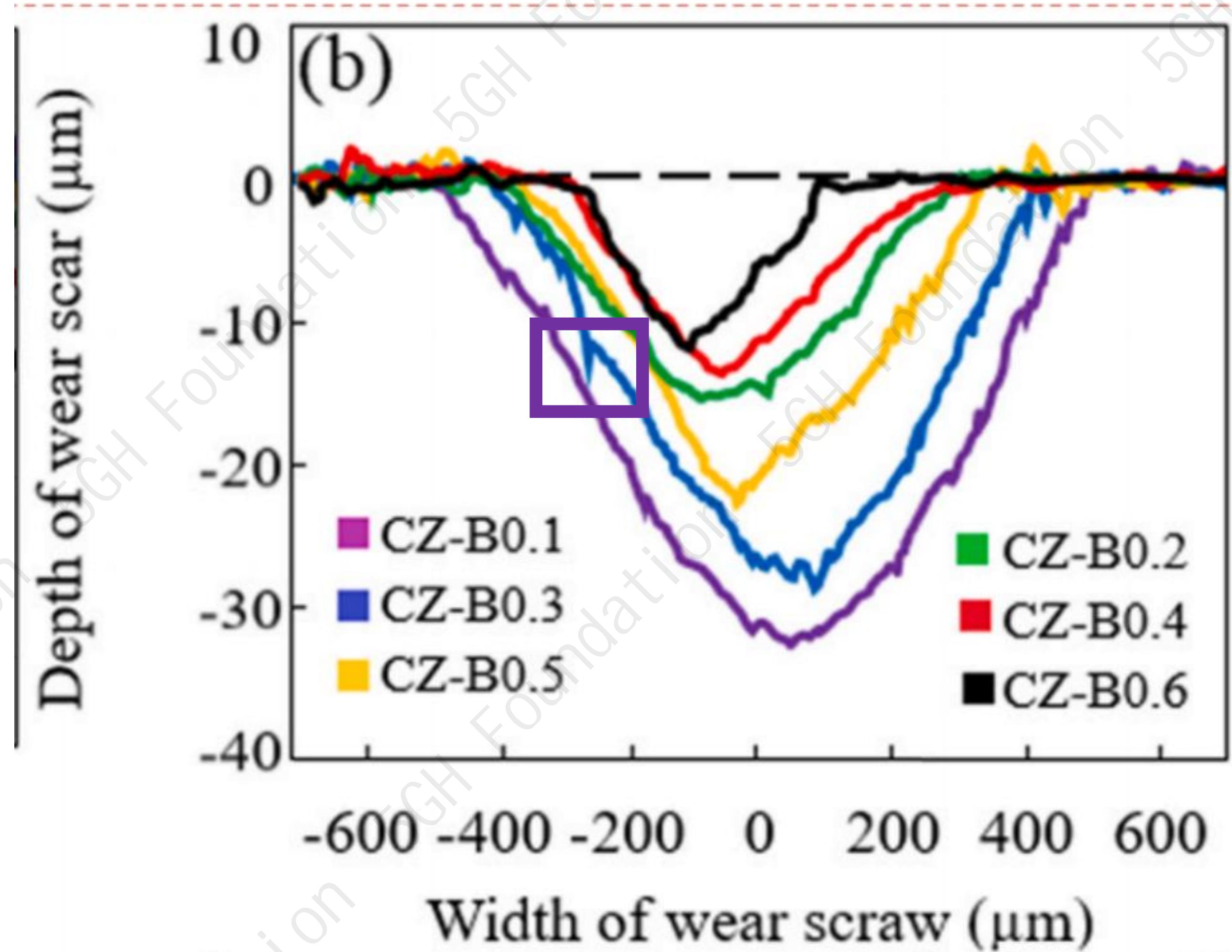
Figure 2(b), XRD



[10.1016/j.matchemphys.2024.129428](https://doi.org/10.1016/j.matchemphys.2024.129428)

Figure 12(b), wear test

Is this kind of “anvil” in purple rectangle normal in wear test?



10.1016/j.matchemphys.2024.128898

Figure 10 and 20, lengths of the error bars seem identical

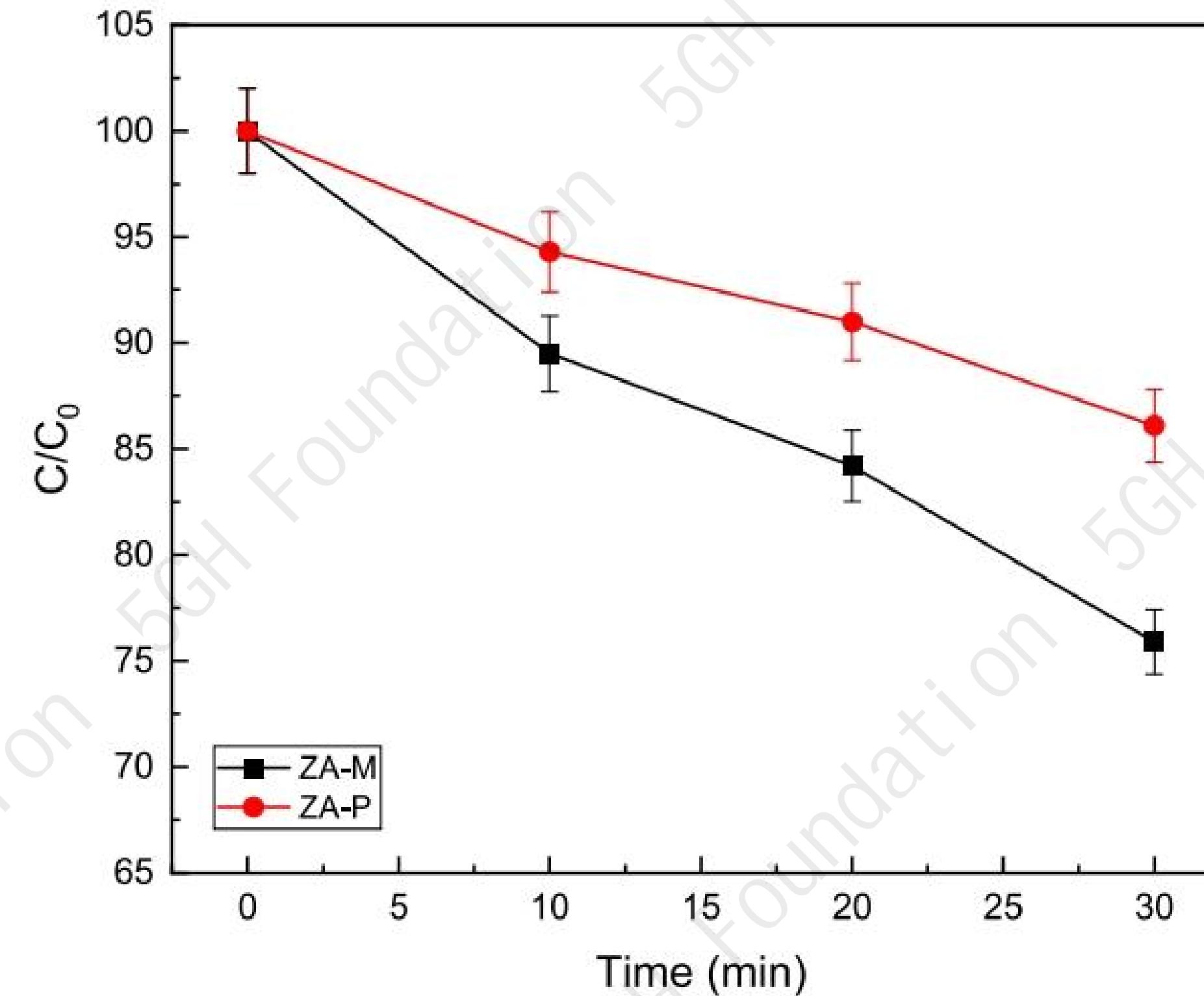


Fig. 10. Removal of MB in dark using ZnAl₂O₄ nanoparticles prepared with different methods.

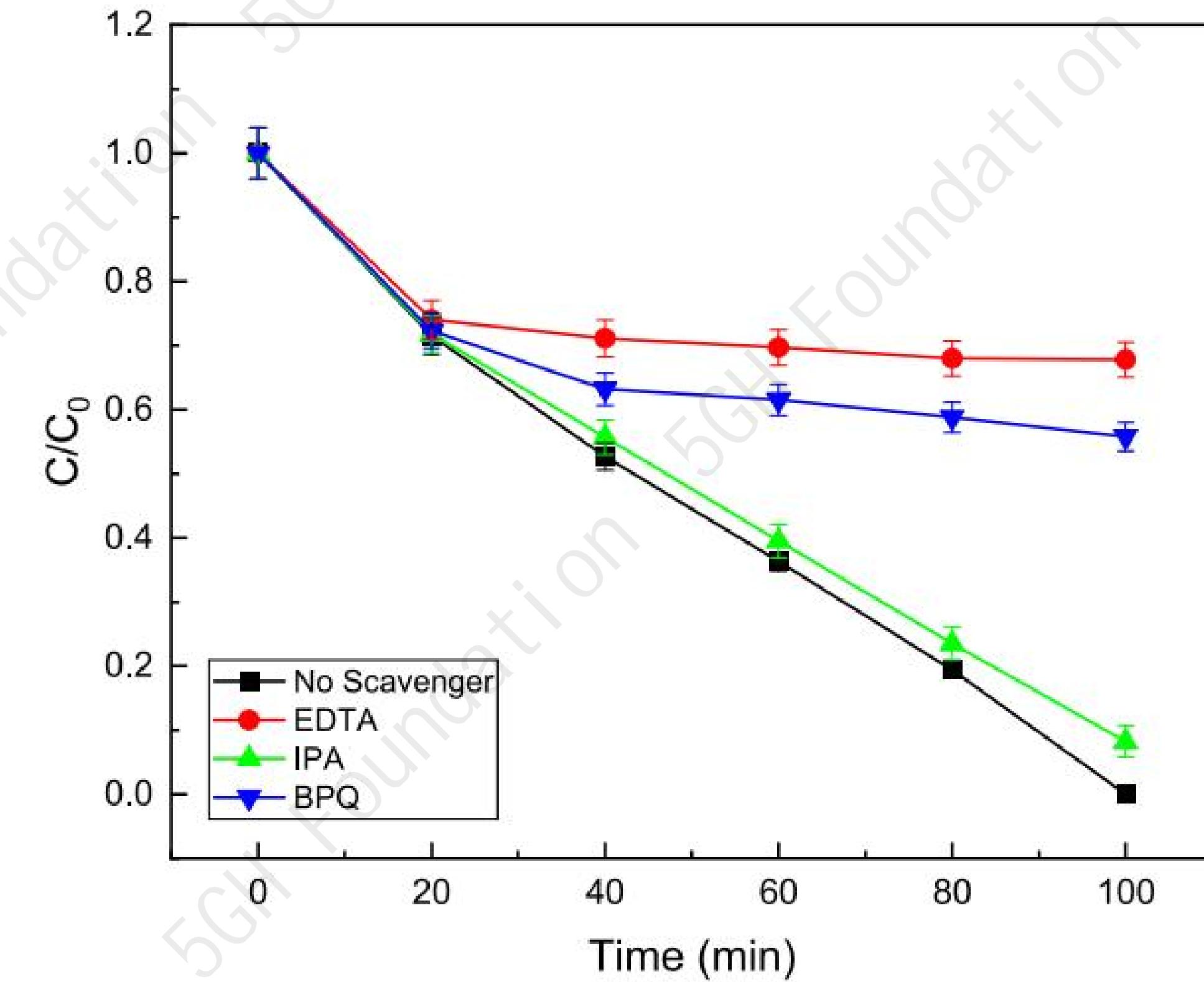


Fig. 20. Effect of different scavengers on the degradation of MB using ZnAl₂O₄-Ag nanoparticles.

10.1016/j.matchemphys.2024.128898

Figure 12, noises for the different lines are identical

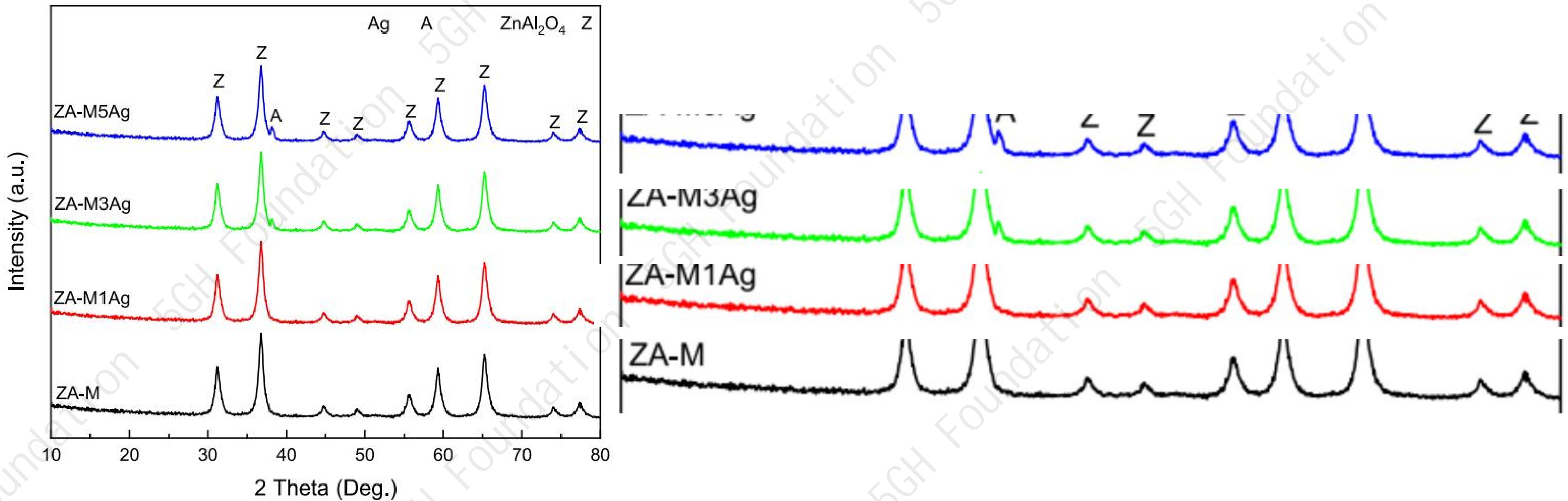


Fig. 12. The X-ray diffraction patterns of synthesized zinc aluminate with amounts of metallic silver.

10.1016/j.matchemphys.2024.128898
Figure 14, lines are identical

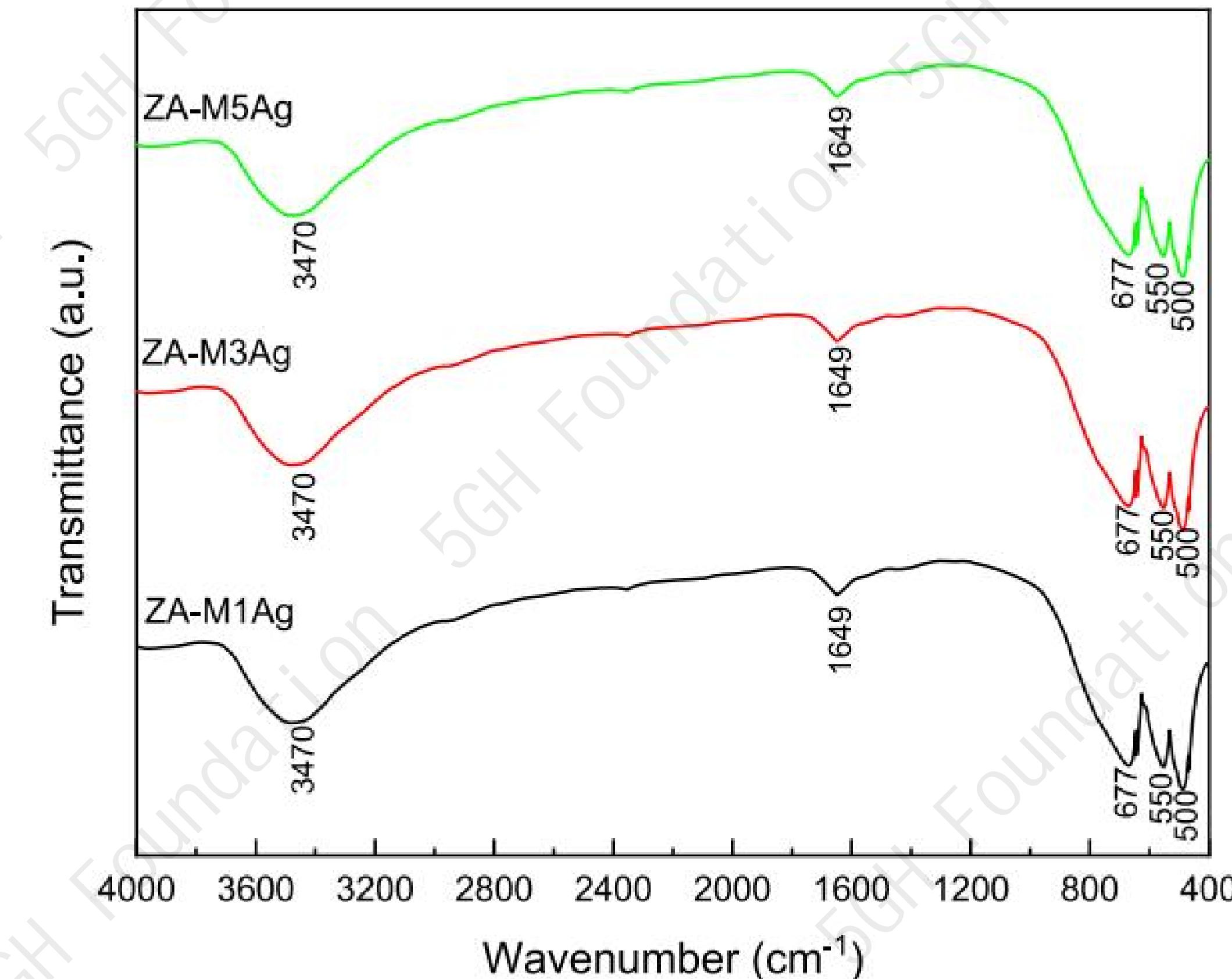


Fig. 14. FTIR analysis of ZnAl₂O₄ nanoparticles with different amounts of metallic silver.

10.1016/j.matchemphys.2024.129876

Table 1: too small standard deviation

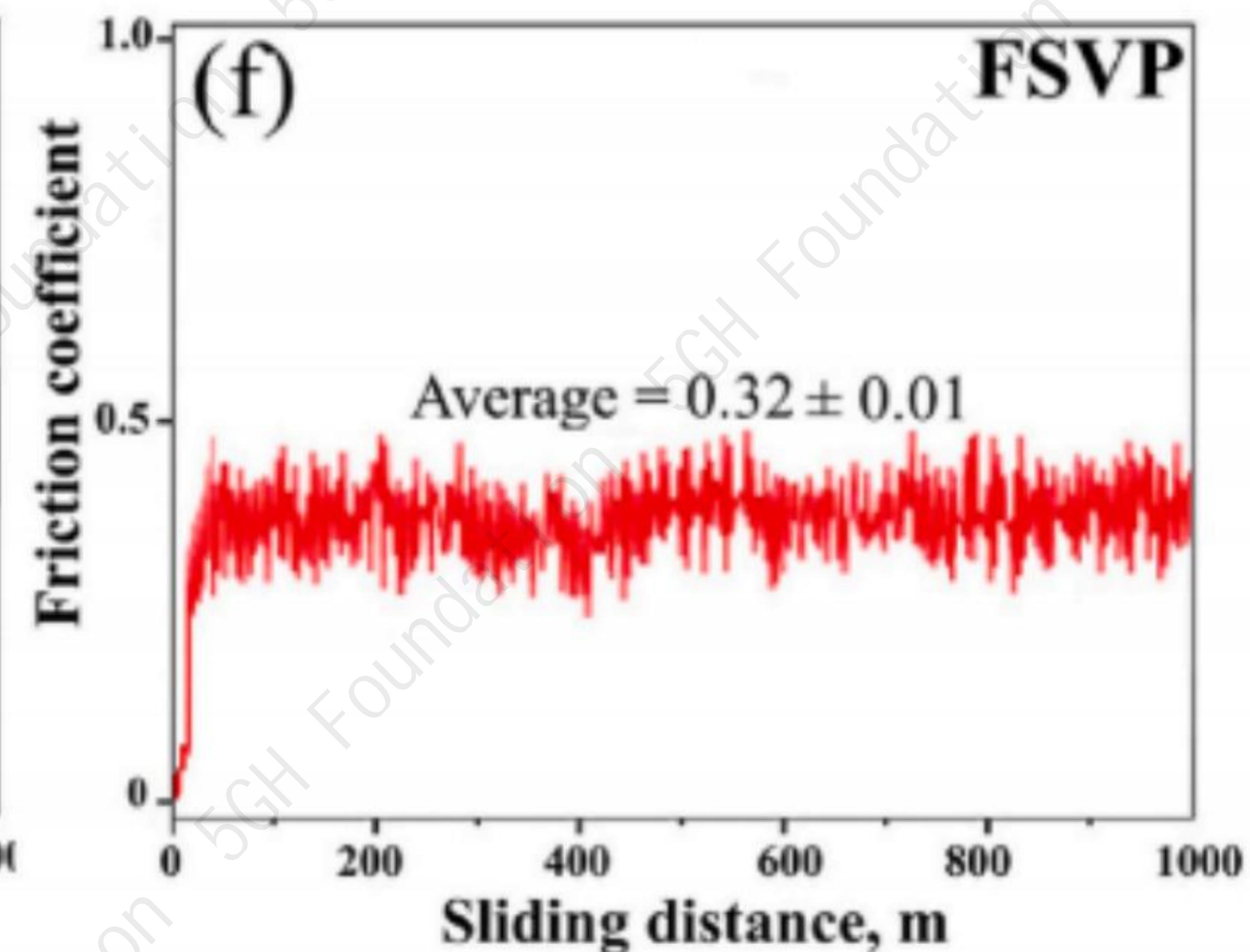
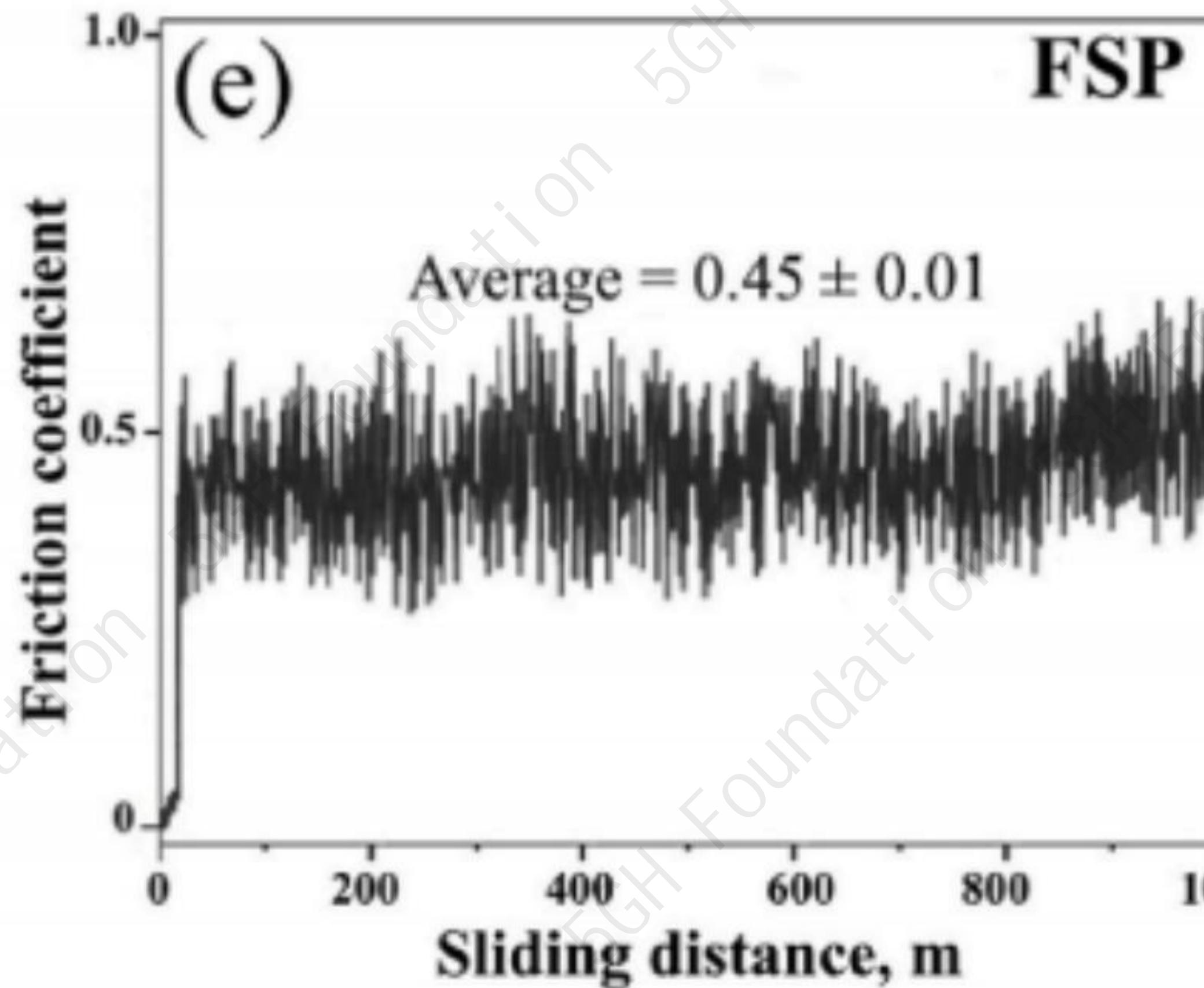
Table 1

Density, bending strength and hardness of the prepared composites.

Sintering method/Temperature	Relative density (%)	Bending strength (MPa)	Hardness (HV1/10)	Average grain size (μm)
Microwave 600 °C	89.12 \pm 0.1	137 \pm 11	78 \pm 7	20.3 \pm 2.8
Microwave 650 °C	92.21 \pm 0.1	162 \pm 10	93 \pm 8	21.9 \pm 2.7
Microwave 700 °C	96.11 \pm 0.1	187 \pm 9	108 \pm 7	21.8 \pm 2.8
SPS 450 °C	99.51 \pm 0.2	231 \pm 11	116 \pm 6	68.4 \pm 9.3

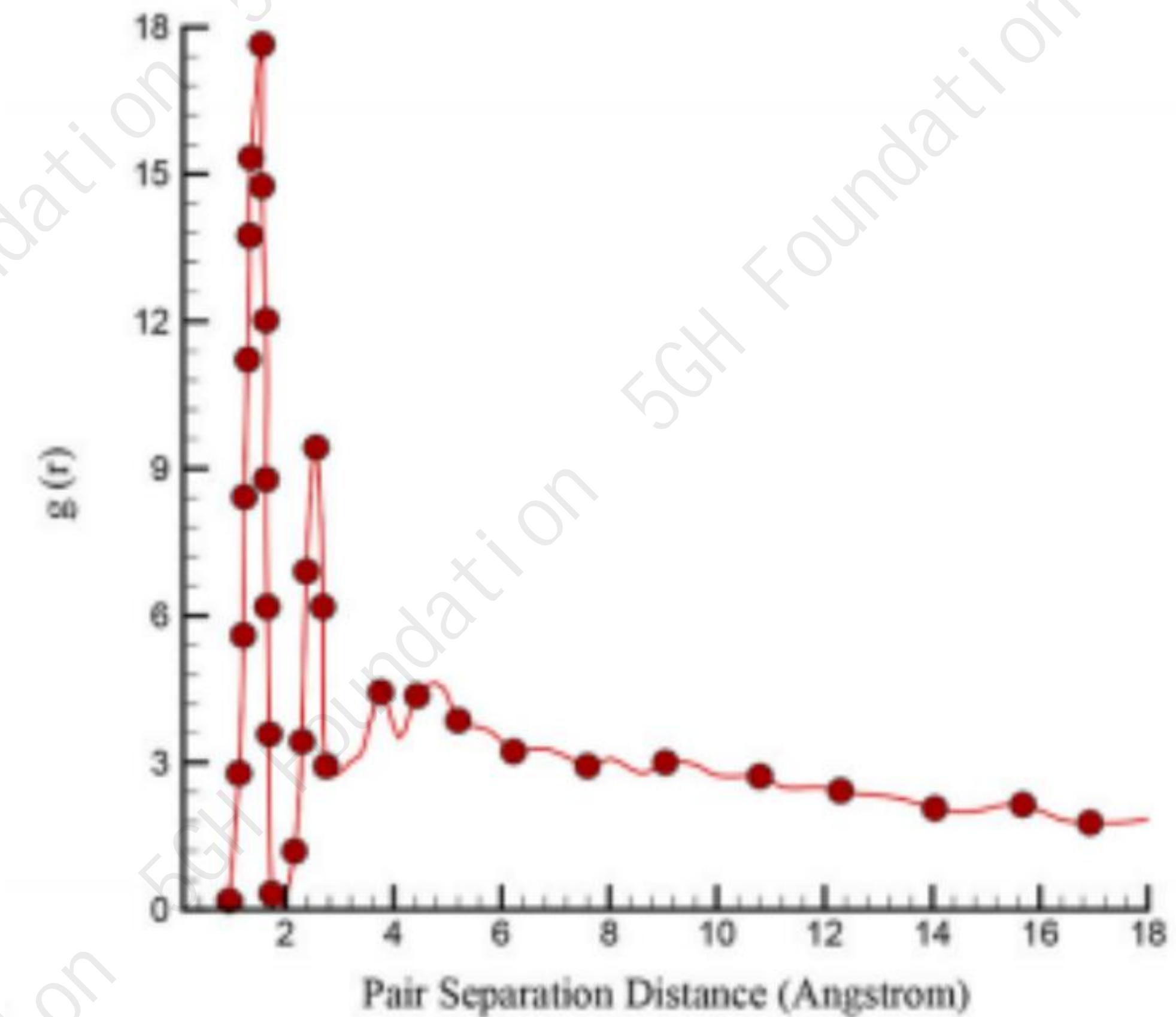
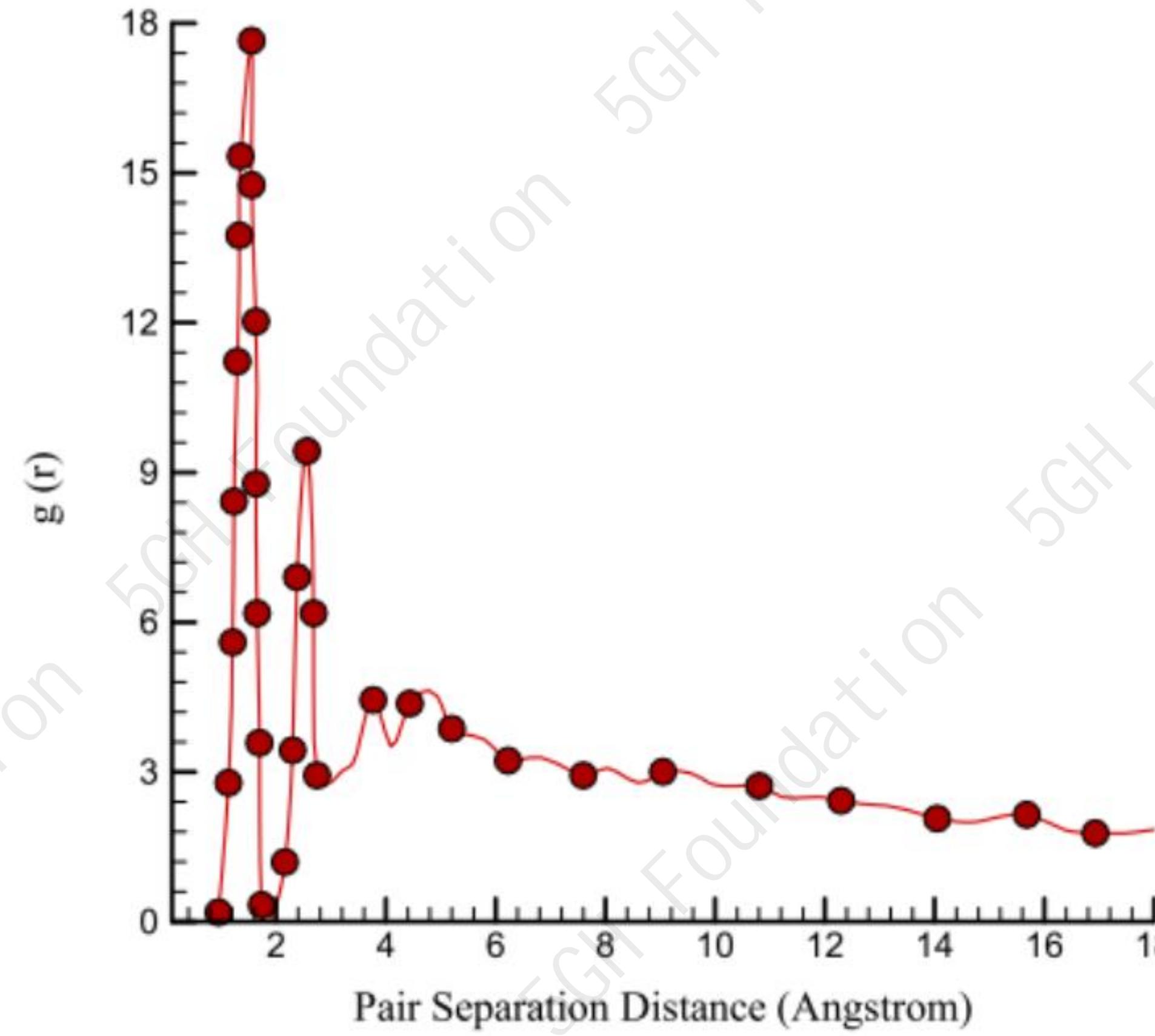
10.1016/j.matchemphys.2024.129149

Figure 6: too small standard deviation



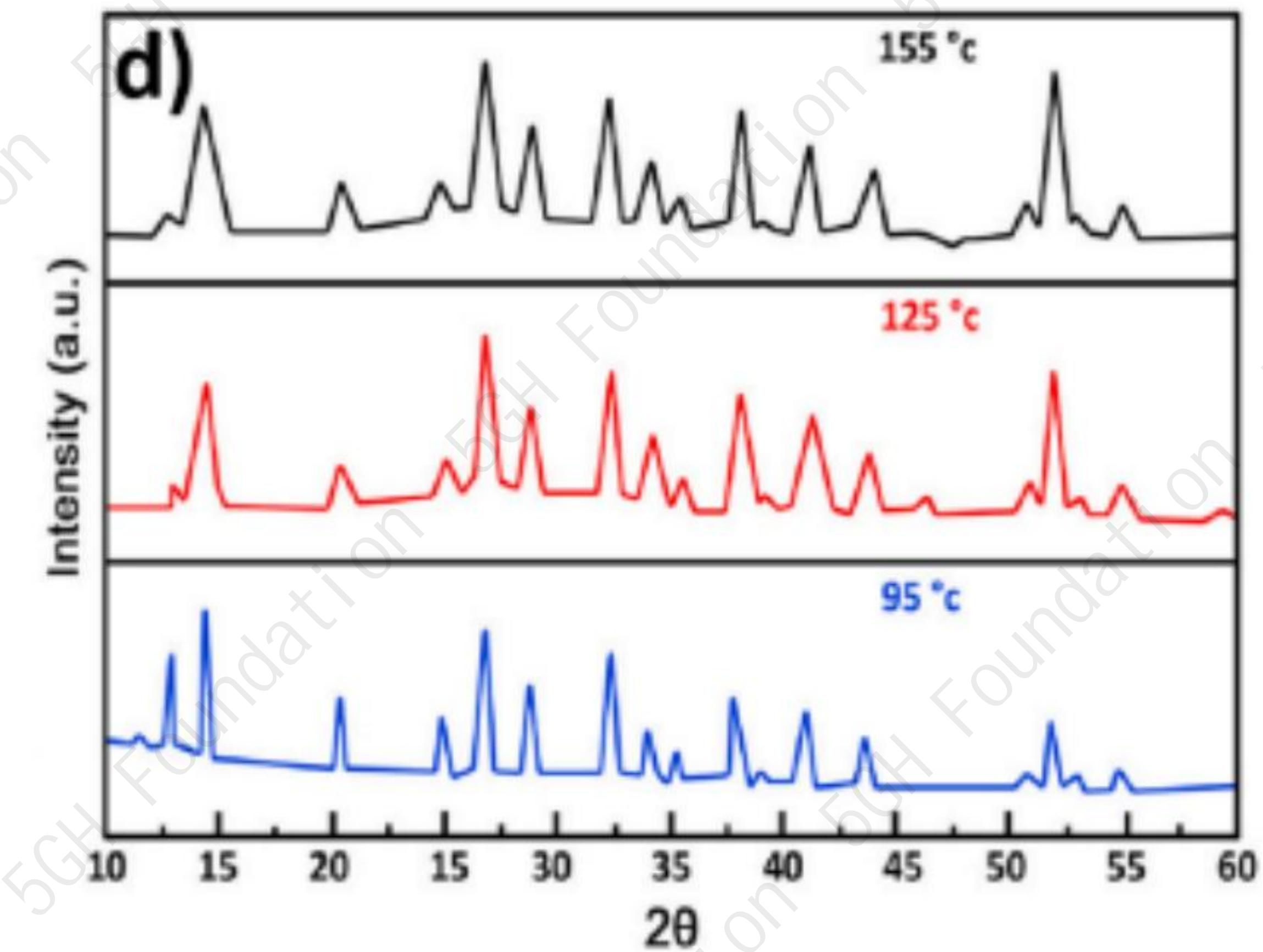
[10.1016/j.matchemphys.2023.127555](https://doi.org/10.1016/j.matchemphys.2023.127555)

Figure 5 (left) and Figure 9 (a) (right) are identical



10.1016/j.matchemphys.2023.127829

Figure 3 (d): abnormal XRD?



10.1016/j.matchemphys.2023.128465

Figure 13 and 14 have same data

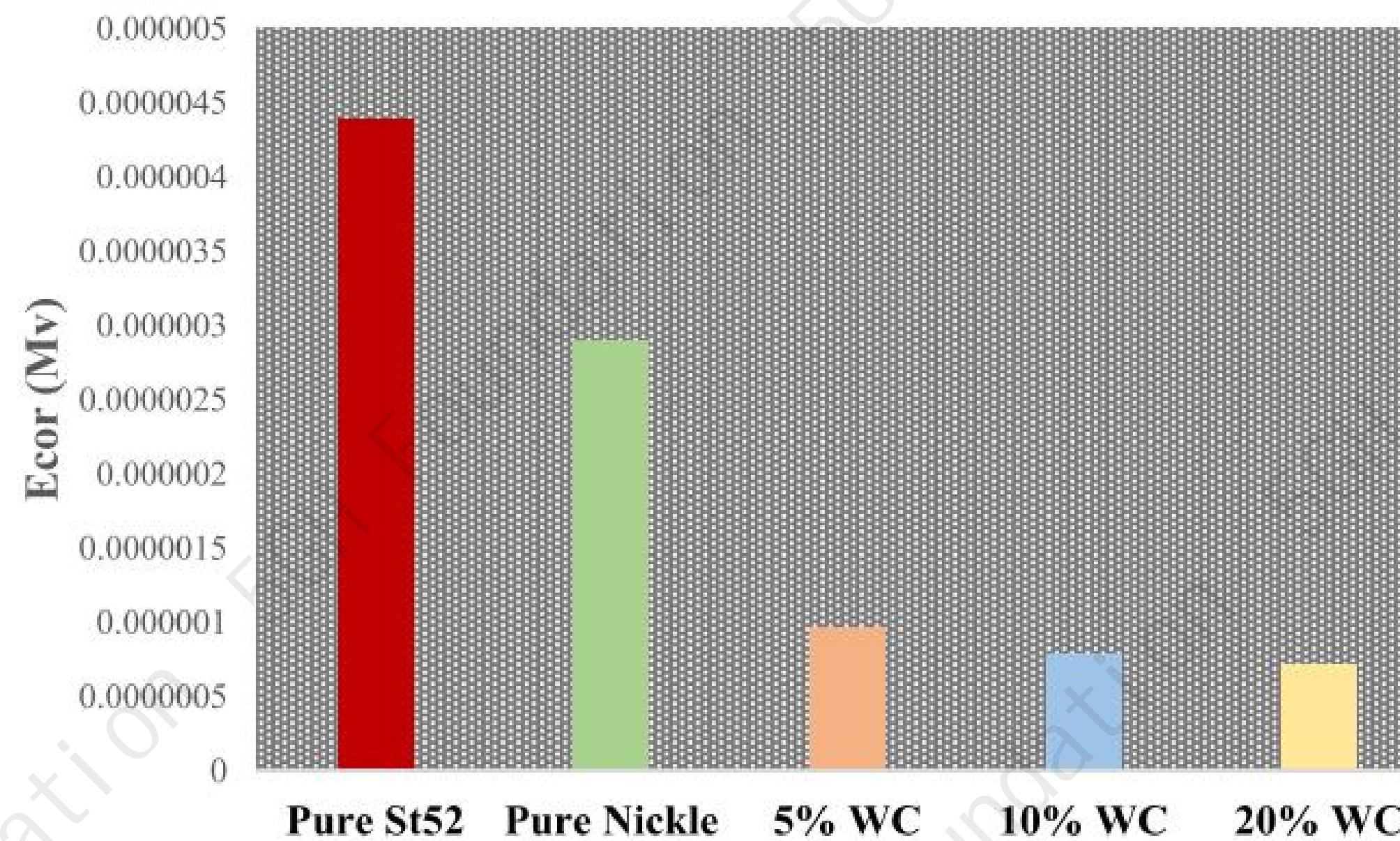


Fig. 13. Changing the corrosion potential of the pure and coated sample with various loading of WC.

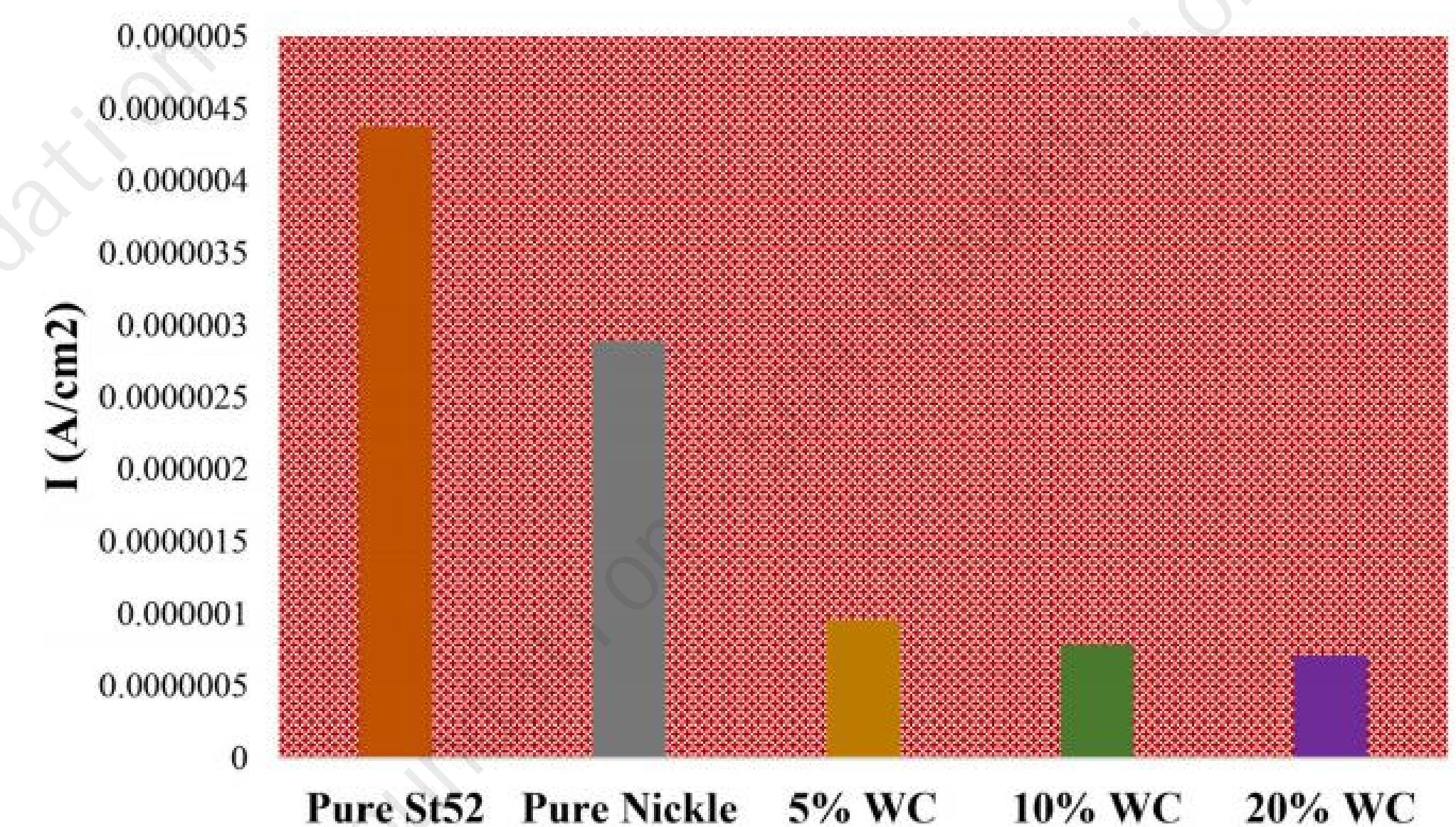


Fig. 14. Changing the corrosion current density in uncoated and coated samples with different percentages of reinforcing particles.

10.1016/j.matchemphys.2023.127855

Figure 1: broken XRD, Figure 5 and 6: error bars with same lengths

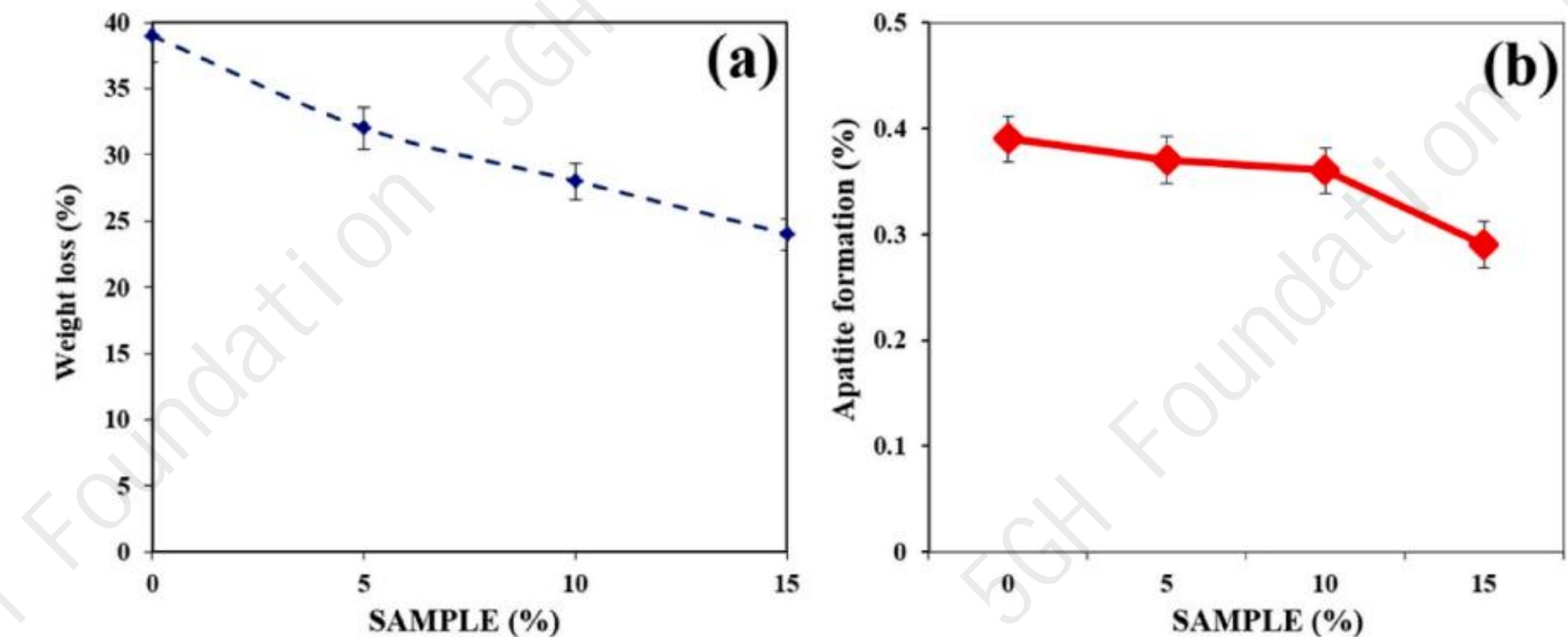
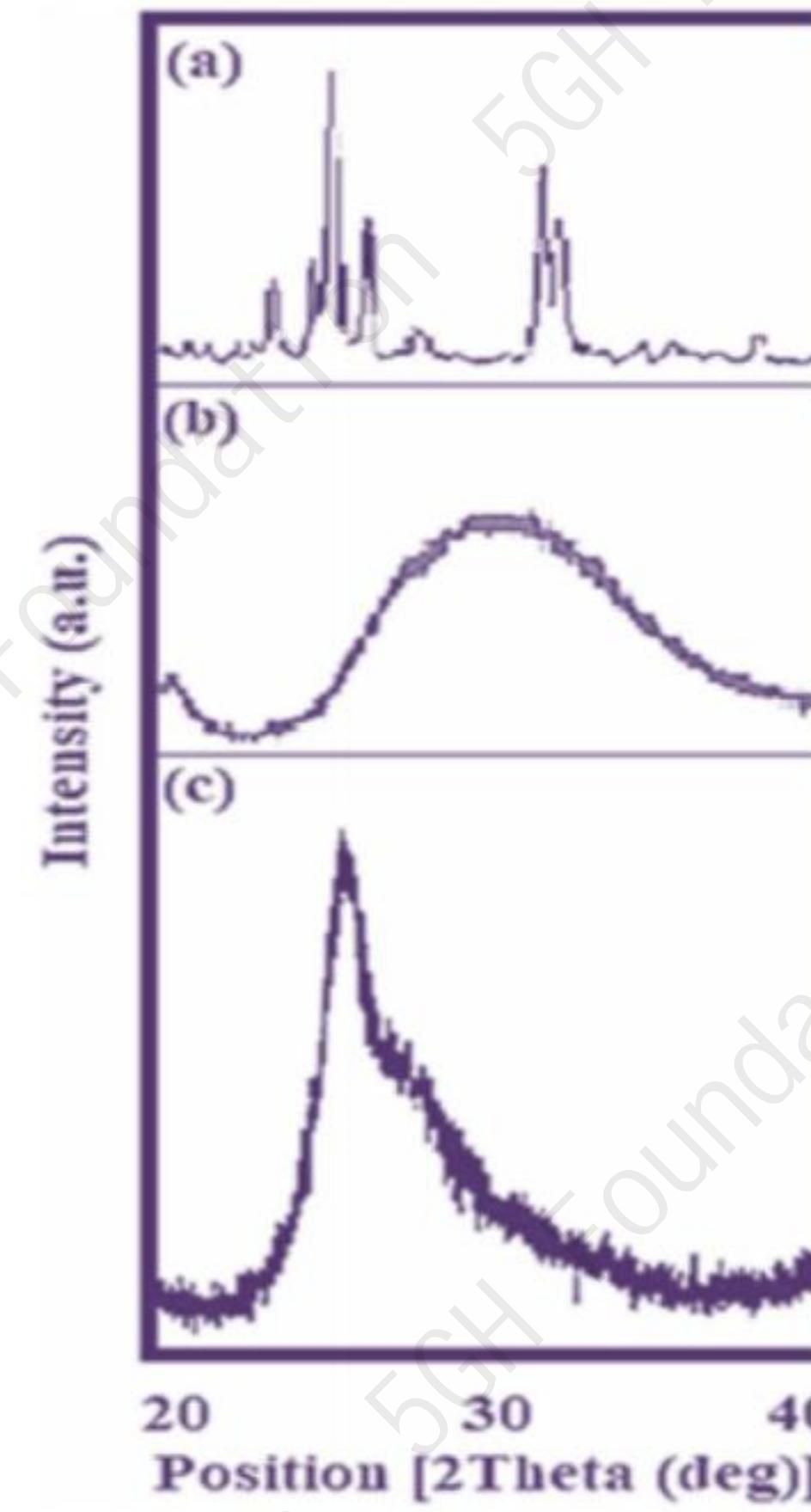


Fig. 5. Evaluation of weight loss and apatite growth results of nanocomposite scaffolds immersed in a SBF after 48 h in water bath.

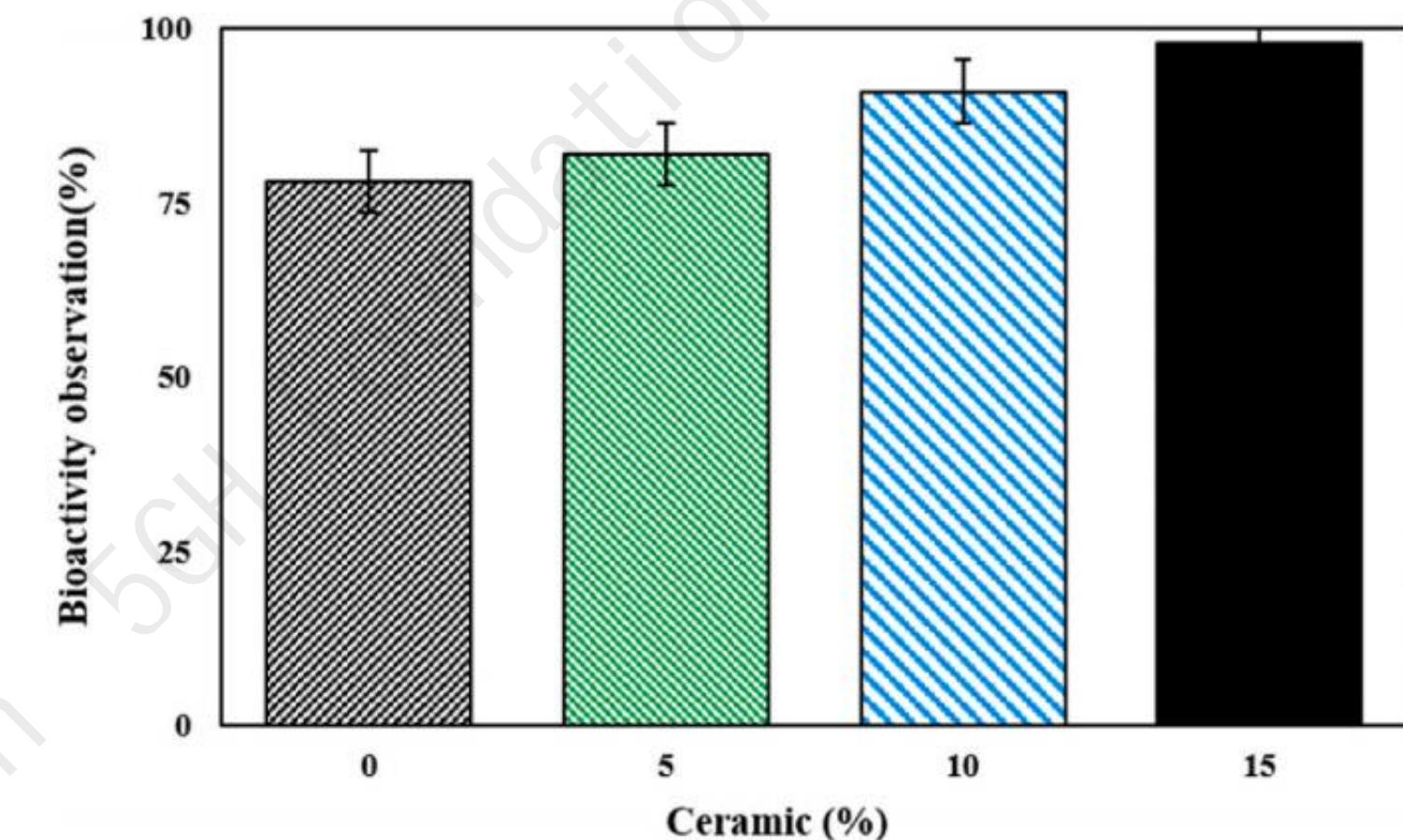
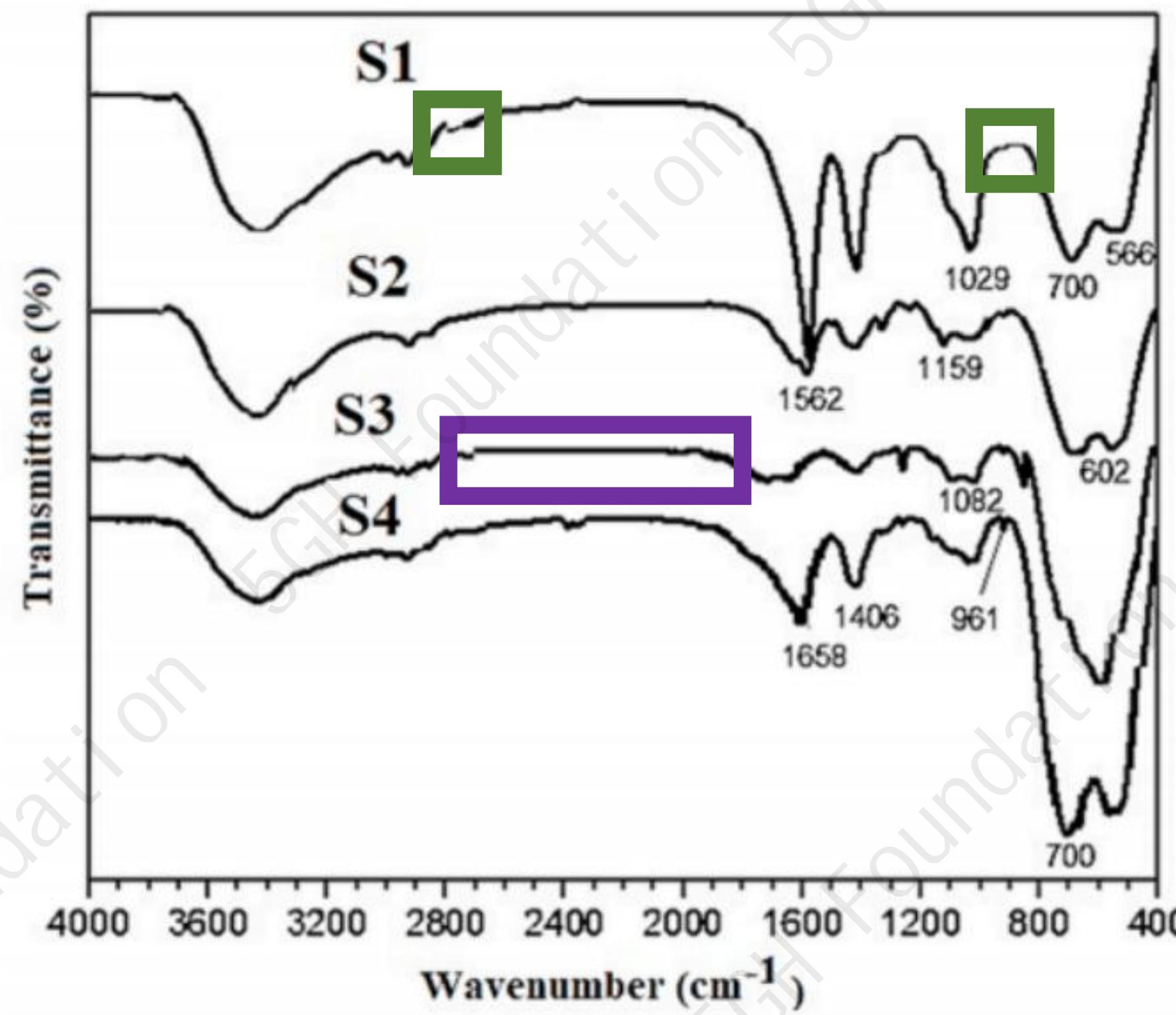


Fig. 6. Results of bioavailability using MTT results of cell growth of porous scaffolds.

10.1016/j.matchemphys.2022.125770, Figure 4



Noise free?

Thinner than
other parts?

Fig. 4. FTIR spectroscopy of wounds containing 0, 5, 10 and 15 wt% of TiO₂.

[10.1016/j.matchemphys.2022.125770](https://doi.org/10.1016/j.matchemphys.2022.125770)

Figure 7: error bars with same lengths

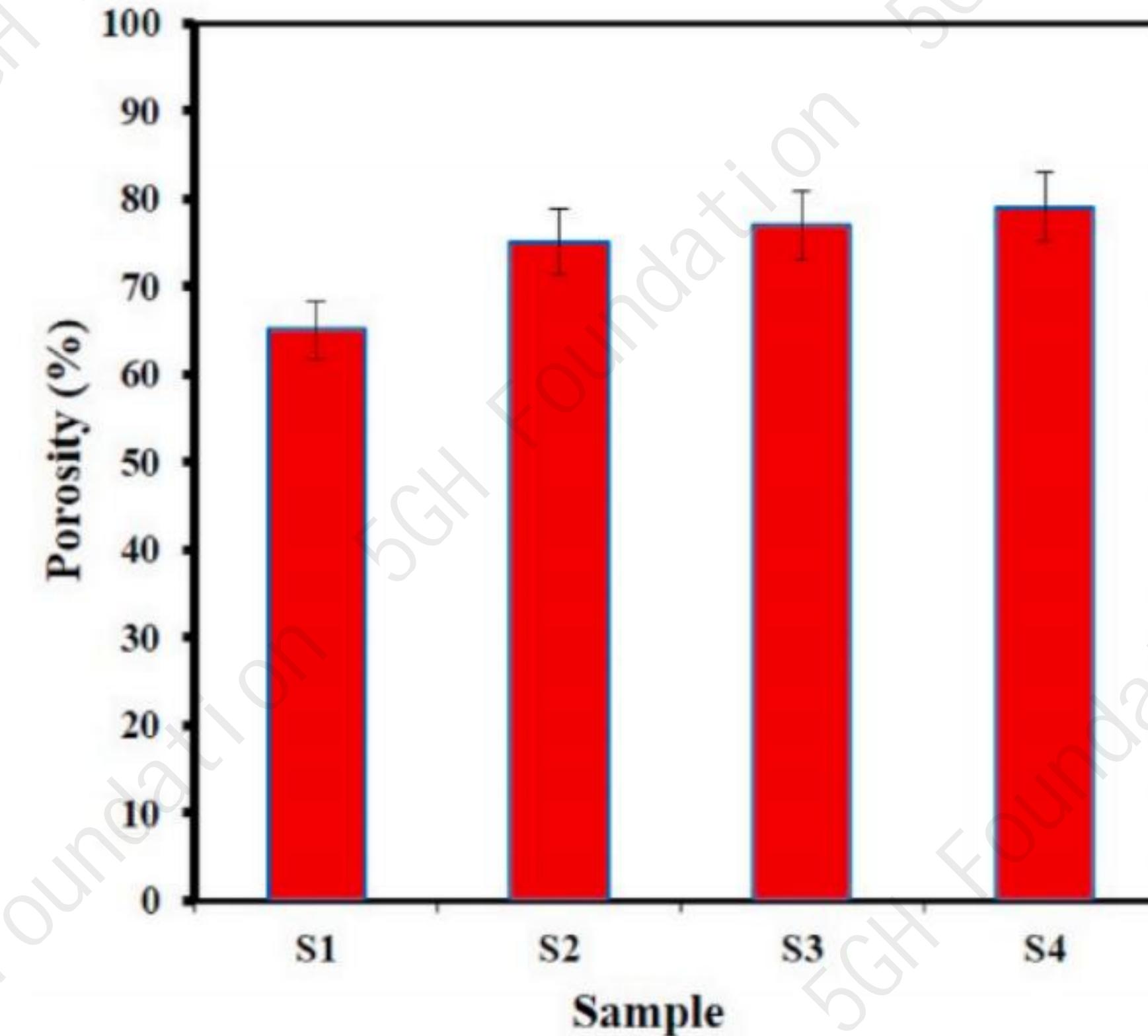


Fig. 7. Porosity percentages of samples containing 0, 5, 10 and 15 wt% of Ago-TiO₂.

10.1016/j.matchemphys.2022.127157

Figure 4: most error bars have same lengths

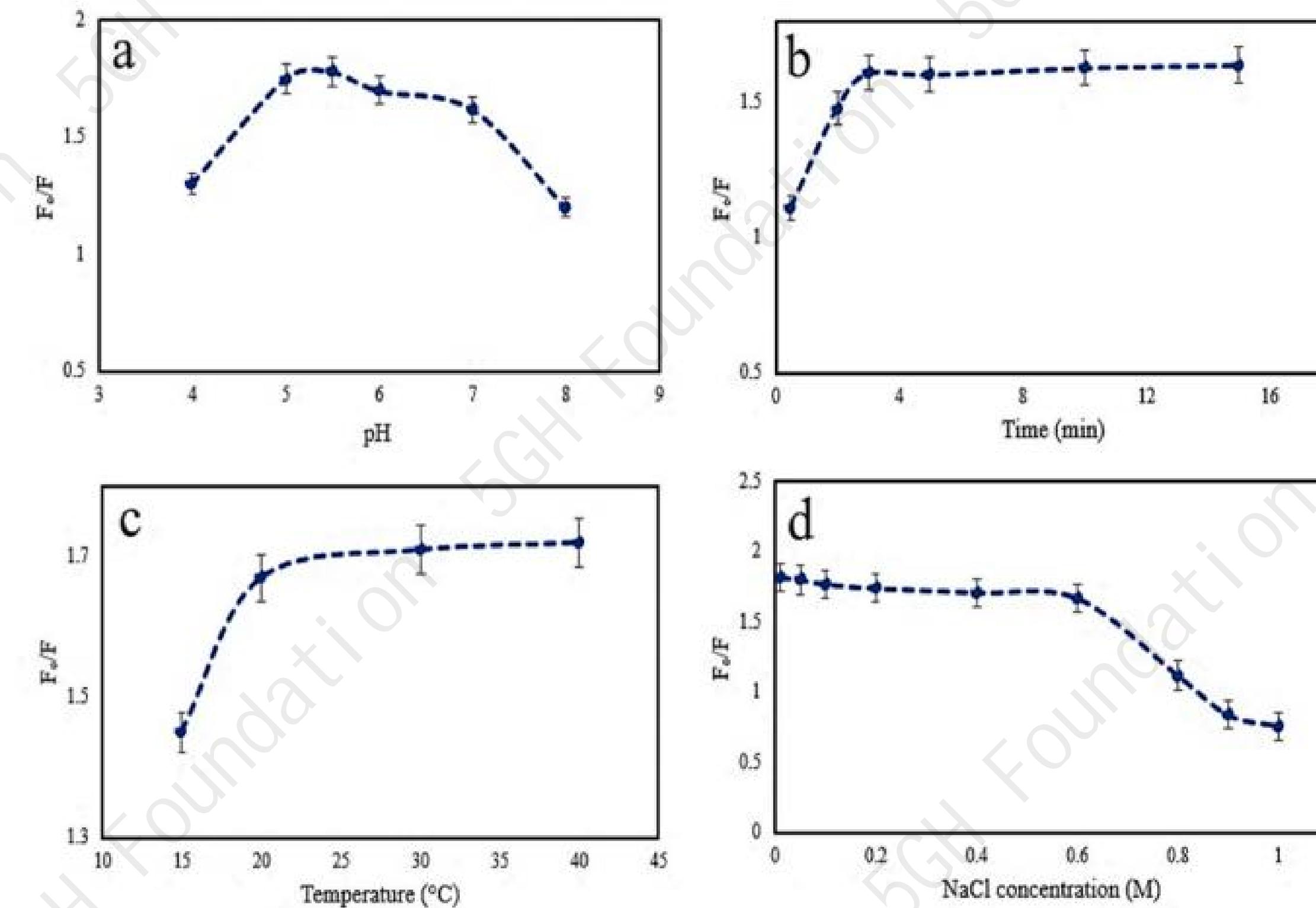


Fig. 4. The effect of (a) pH (Concentration: 30 μ M, time: 5 min, temperature: 25 °C), (b) time (Concentration: 30 μ M, pH: 5.5, temperature: 25 °C) (c) temperature (Concentration 30 μ M, pH: 5.5, time: 3 min)and (d) ionic strength (Concentration: 30 μ M, pH: 5.5, Time:3 min, temperature:25 °C) on the FL quenching response of HNT/MOF (IMA concentration, 30 μ M). Results are the average of three replicates measurements.

[10.1016/j.matchemphys.2021.124720](https://doi.org/10.1016/j.matchemphys.2021.124720)

Figure 2: most parts in the upper two lines are identical

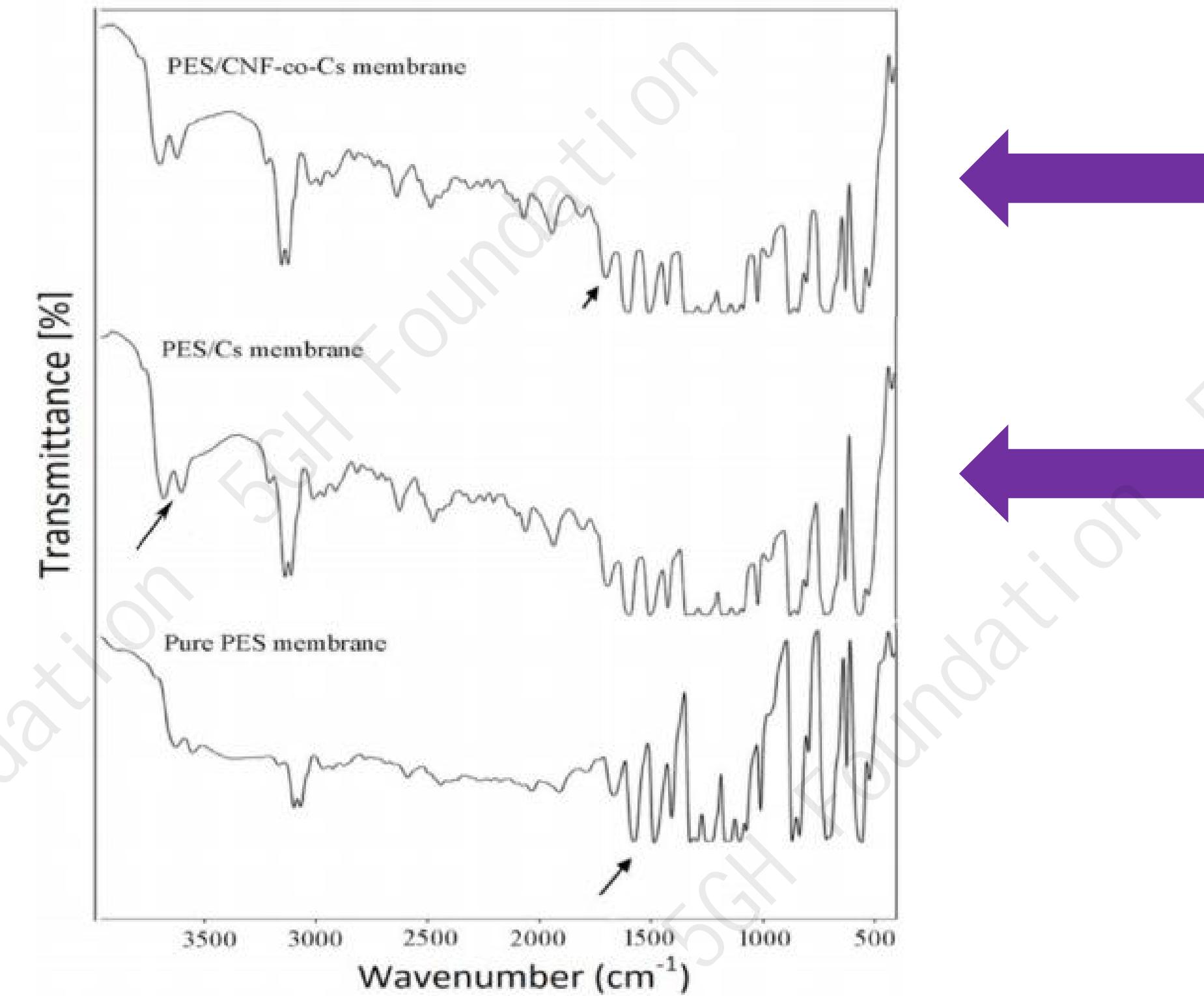


Fig. 2. The FTIR spectrum of PES membrane, PES/Cs and PES/CNFs-co-Cs membranes.

10.1016/j.matchemphys.2021.124990

Figure 1: seems weird

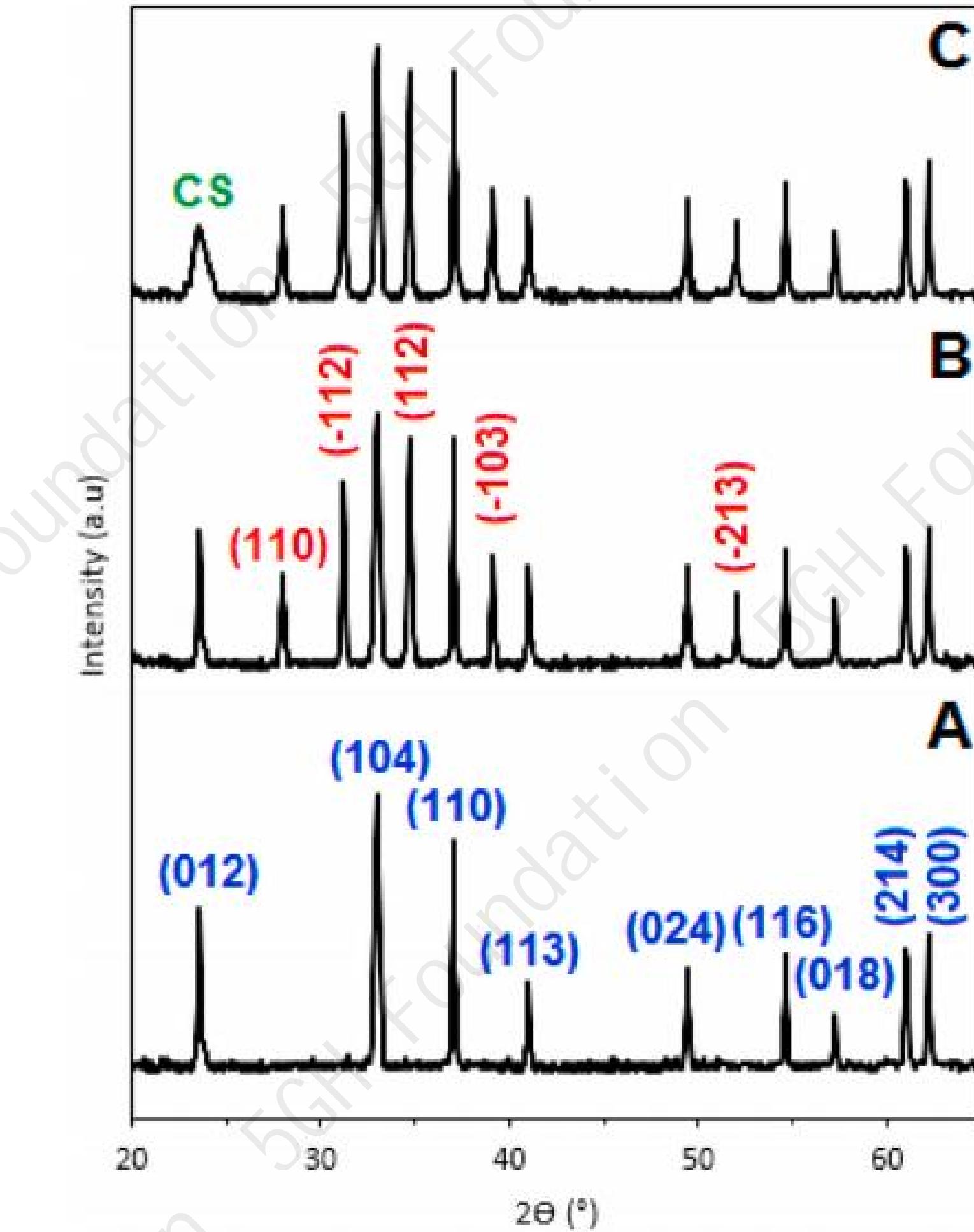


Fig. 1. XRD curves of NTO (A), AS/NTO (B), and AS/NTO/CS (C) nanocomposites.

10.1016/j.matchemphys.2011.04.083

Table 1 and 2: same data for water?

Table 1

T1 relaxivity measurements for 4 concentration of dextran coated USPIO (sample D) and samples A, B and C.

mg ml ⁻¹	0.015	0.008	0.004	0.002	Water
Sample A T1	257.3	483.1	1008.9	2130.4	2248.2
Sample B T1	110.4	183.0	302.1	497.5	2248.2
Sample C T1	125.4	234.5	420.9	891.3	2248.2
Sample D T1	253.0	527.1	102.7	2193.0	2248.2

Table 2

T2 relaxivity measurements for 4 concentration of dextran coated USPIO (sample D) and samples A, B and C.

mg ml ⁻¹	0.015	0.008	0.004	0.002	Water
Sample A T2	19.8	23.0	55.2	201.2	389.9
Sample B T2	18.9	22.9	39.3	67.8	389.9
Sample C T2	18.0	30.4	58.8	112.7	389.9
Sample D T2	22.1	24.7	50.8	150.7	389.9

10.1016/j.matchemphys.2019.122233

Table 2: same standard deviation

Table 2
Roughness parameters, coating thickness, and critical load of $\text{TiO}_2\text{-Al}_2\text{O}_3\text{-BTA}$ and FG coatings.

Parameters	Coatings		Al2024
	TiO ₂ -Al ₂ O ₃ -BTA coating	FG coating	
Surface Roughness (nm)	4.1	4.4	217
Coating Thickness of each layer (nm)	Titania Ti-Al- 2.4	$300 \pm 10 \text{ nm}$ $2280 \pm 10 \text{ nm}$	Titania Ti-Al- 1.2 2.4 Ti-Al- 3.6 Ti-Al- 4.8
Total Coating Thickness (nm)		$2580 \pm 10 \text{ nm}$	$300 \pm 10 \text{ nm}$ $373 \pm 10 \text{ nm}$ $575 \pm 10 \text{ nm}$ $592 \pm 10 \text{ nm}$ $610 \pm 10 \text{ nm}$
Critical load (mN)	17	18	-

10.1016/j.matchemphys.2019.122233

Table 3: too small and same standard deviation

Table 3

Polarization analysis parameters and released Al after immersion for 1 h and 96 h in 3.5 wt% NaCl.

Sample	Immersion time (h)	(mV)	(mA/cm ²)	(mV/decade)	β_a (mV/decade)	Corrosion Rate (mm/year)	Al (mg/L.cm ²)	Efficiency (%)
AA2024-T3		-693 ± 1	15E-4	24	16	1.92E-3	3.6471	-
TiO ₂ -Al ₂ O ₃ -BTA	1	-862 ± 1	26E-6	42	77	5.75E-05	0.0142	97
FG coating		-824 ± 1	14E-6	38	13	2.45E-05	0.0037	98.6
AA2024-T3		-761 ± 1	17E-3	56	89	4.32E-04	8.5792	-
TiO ₂ -Al ₂ O ₃ -BTA	96	-753 ± 1	35E-6	25	22	3.12E-05	0.0836	92.8
FG coating		-713 ± 1	65E-7	47	73	2.32E-05	0.0059	94.5